

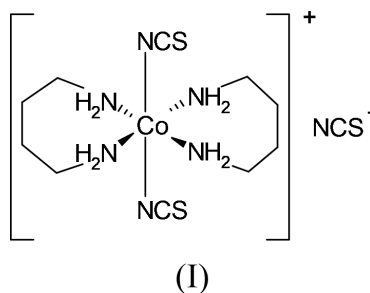
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## Key indicators

Single-crystal X-ray study  
 $T = 296$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å  
 $R$  factor = 0.033  
 $wR$  factor = 0.094  
Data-to-parameter ratio = 27.0For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.*trans*-Bis(1,4-butanediamine)bis(isothiocyanato)-cobalt(III) thiocyanateThe title compound,  $[\text{Co}(\text{NCS})_2(\text{C}_4\text{H}_{12}\text{N}_2)_2]\text{NCS}$ , (I), has an octahedral coordination geometry, in which one of the seven-membered chelate rings adopts the chair form and the other the twist-boat form.

## Comment

Metal complexes of 1,4-butanediamine, also known as putrescin or tetramethylenediamine (tmd), are of interest in the context of bioinorganic chemistry (Gasowska *et al.*, 2000). Only a few structures have been reported for transition metal complexes of tmd (Sato *et al.*, 1974; Shimoi *et al.*, 1988; Kurachi & Ohba, 1992).

The two geometrical isomers of  $[\text{Co}^{\text{III}}(\text{NCS})_2(\text{tmd})_2]^+$  have been prepared and their geometrical configurations have been determined from their spectroscopic properties (Nagata & Kanamori, 2001). The present X-ray analysis confirms the *trans* configuration for one of the isomers, (I) (Fig. 1). In (I), the two seven-membered chelate rings adopt different conformations (Fig. 2). The conformation of Fig. 2(a) corresponds to the chair form that has been found in *trans*- $[\text{Co}(\text{III})(\text{NO}_2)_2(\text{tmd})_2]^+$  (Shimoi *et al.*, 1988). The other chelate ring (Fig. 2b) adopts approximately the twist-boat form. The coordination bond distances and angles are in the normal ranges.

## Experimental

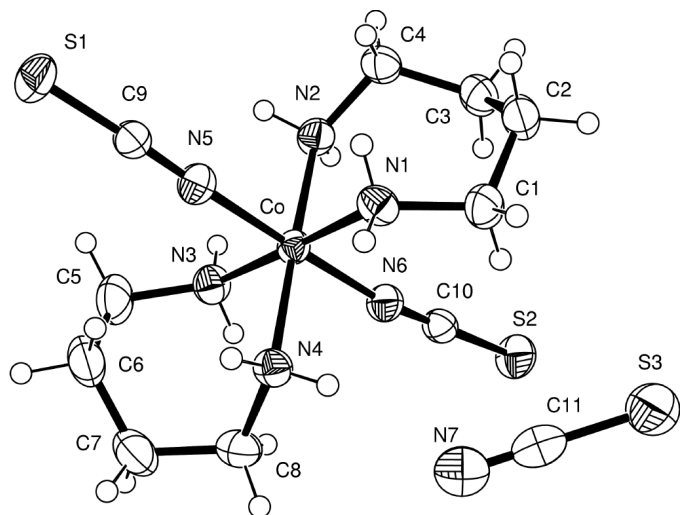
The title compound, (I), was prepared by adapting the procedures described by Nagata *et al.* (1985) and Nagata & Kanamori (2001). Crystals of (I) were obtained by evaporation of an aqueous solution at room temperature.

## Crystal data

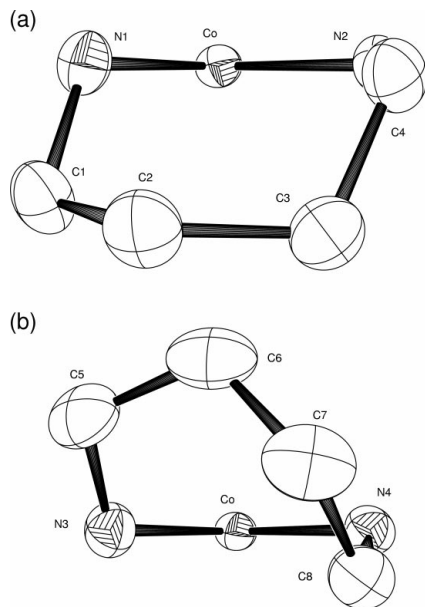
$[\text{Co}(\text{NCS})_2(\text{C}_4\text{H}_{12}\text{N}_2)_2]\text{NCS}$   
 $M_r = 409.48$   
 Monoclinic,  $P2_1/c$   
 $a = 9.739$  (2) Å  
 $b = 13.647$  (4) Å  
 $c = 14.5479$  (19) Å  
 $\beta = 107.790$  (13)°  
 $V = 1841.2$  (7) Å<sup>3</sup>  
 $Z = 4$

$D_x = 1.477$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 Cell parameters from 25 reflections  
 $\theta = 14.8$ – $15.0$ °  
 $\mu = 1.28$  mm<sup>-1</sup>  
 $T = 296$  (2) K  
 Prism, brown  
 $0.20 \times 0.20 \times 0.15$  mm

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**Figure 1**  
The molecular structure of (I) showing the labelling of the non-H atoms. Displacement ellipsoids are shown at 50% probability levels.



**Figure 2**  
The conformations of the chelate rings projected along the bisector of the N—Co—N angle.

#### Data collection

Rigaku AFC-7R diffractometer  
 $\omega$ - $2\theta$  scans  
Absorption correction:  $\psi$  scan  
(North *et al.*, 1968)  
 $T_{\min} = 0.784$ ,  $T_{\max} = 0.831$   
5653 measured reflections  
5371 independent reflections  
4027 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.027$   
 $\theta_{\text{max}} = 30.0^\circ$   
 $h = 0 \rightarrow 13$   
 $k = 0 \rightarrow 19$   
 $l = -20 \rightarrow 19$   
3 standard reflections  
frequency: 60 min  
intensity decay: 1.8%

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.033$   
 $wR(F^2) = 0.094$   
 $S = 1.06$   
5371 reflections  
199 parameters  
H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0409P)^2 + 0.8638P]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\text{max}} = 0.001$$

$$\Delta\rho_{\text{max}} = 0.42 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\text{min}} = -0.56 \text{ e } \text{\AA}^{-3}$$

**Table 1**

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

Co—N1	1.9881 (17)	Co—N4	1.9930 (17)
Co—N2	1.9747 (17)	Co—N5	1.8918 (17)
Co—N3	1.9856 (17)	Co—N6	1.8961 (17)
N1—Co—N2	92.21 (7)	N2—Co—N6	89.10 (7)
N1—Co—N3	177.26 (7)	N3—Co—N4	90.44 (7)
N1—Co—N4	88.51 (7)	N3—Co—N5	92.13 (8)
N1—Co—N5	85.41 (8)	N3—Co—N6	87.45 (8)
N1—Co—N6	95.05 (7)	N4—Co—N5	93.29 (8)
N2—Co—N3	88.94 (7)	N4—Co—N6	88.60 (7)
N2—Co—N4	177.64 (7)	N5—Co—N6	178.07 (8)
N2—Co—N5	89.01 (8)		

H atoms bonded to C and N atoms were placed geometrically and refined using a riding model *via* the *SHELXL97 HFIX/AFIX 23* facility. The displacement parameter was set as 1.2 times that of the parent atom.

Data collection: *AFC-7R Diffractometer Control Software* (Rigaku, 1999); cell refinement: *AFC-7R Diffractometer Control Software*; data reduction: *AFC-7R Diffractometer Control Software*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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#### References

- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.  
Gasowska, A., Lomozik, L. & Jastrzab, R. (2000). *J. Inorg. Biochem.* **78**, 139–147.  
Kurachi, S. & Ohba, S. (1992). *Bull. Chem. Soc. Jpn.* **65**, 3033–3041.  
Nagata, K. & Kanamori, K. (2001). *J. Coord. Chem.* In the press.  
Nagata, K., Kanamori, K., Kawai, K. & Ogino, H. (1985). *Chem. Lett.* pp. 1507–1510.  
North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.  
Rigaku (1999). *AFC-7R Diffractometer Control Software*. Rigaku Corporation, Tokyo, Japan.  
Sato, S., Saito, Y., Fujita, J. & Ogino, H. (1974). *Inorg. Nucl. Chem. Lett.* **10**, 669.  
Sheldrick, G. M. (1997). *SHELXL97* and *SHELXS97*. University of Göttingen, Germany.  
Shimoi, M., Fujinawa, Y., Ogino, H., Kanamori, K. & Kawai, K. (1988). *Bull. Chem. Soc. Jpn.* **61**, 3491–3496.